MkIV ground-based measurements of Ethene (Ethylene; C₂H₄)

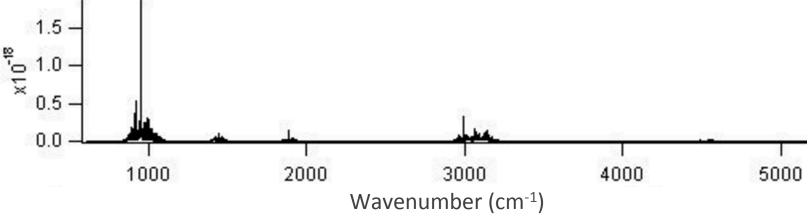


Geoffrey Toon, Jean Francois Blavier, Keeyoon Sung Jet Propulsion Laboratory, California Institute of Technology 2016-06-02

 C_2H_4 has been previously measured by ACE (e.g., Herbin et al., 2009) and from the ground in fires (Paton-Walsh et al., 2005; Rinsland et al., 2005).

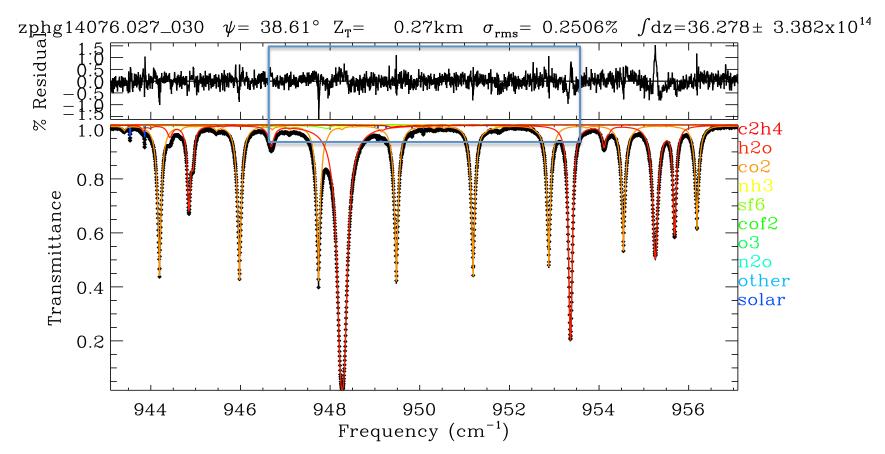
Best IR feature of C₂H₄ is Q-branch of nu₇ band (CH₂ wag) at 948 cm⁻¹. This is 7 times stronger than any feature in the 3000 cm⁻¹ region.

Ethylene images from PNNL



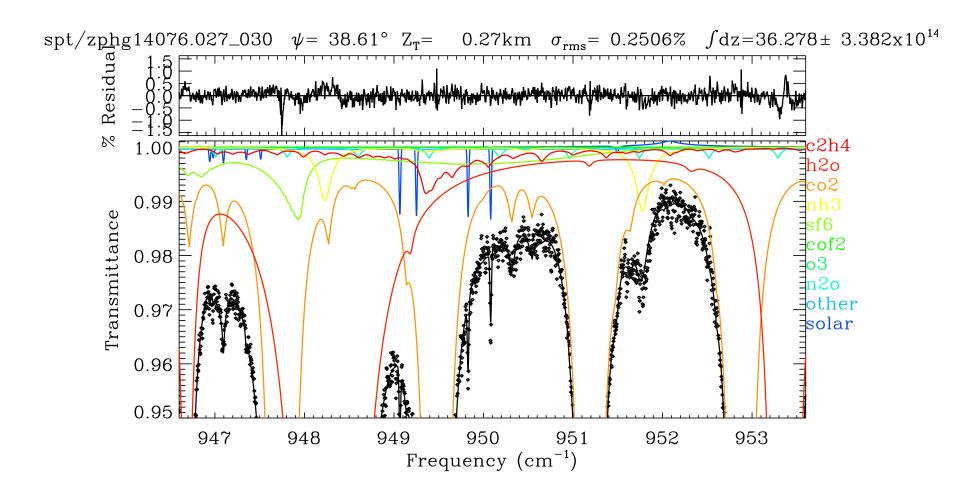
Copyright 2016 Caltech. Government sponsorship acknowledged

Example of Fit to MkIV Spectrum from JPL



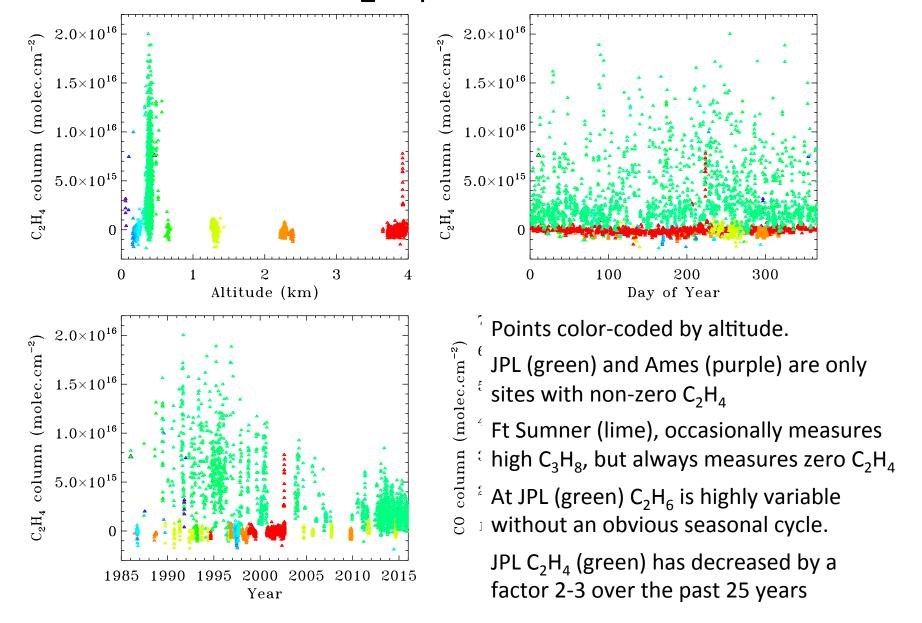
Defined a 14 cm⁻¹ wide window which include several strong lines of CO_2 and H_2O . CO_2 lines are T-dependent, one lies directly overlies C_2H_4 Q-branch. Fitting done using GFIT, NCEP model, ATM linelist (i.e., TCCON methodology). C_2H_4 Q-branch at 949.4 cm⁻¹ < 1% deep and therefore not discernable on this plot.

Zoom into Spectral Fit



 C_2H_4 Q-branch (red) peaks at 949.4 cm⁻¹. It is typically less than 1% deep even at JPL. It lies under a 60%-deep T-dependent CO_2 line centered at 949.5 cm⁻¹, the wing of a saturated H_2O line centered at 948.3 cm⁻¹, and the SF_6 R-branch.

Retrieved MkIV C₂H₄ columns from 12 sites

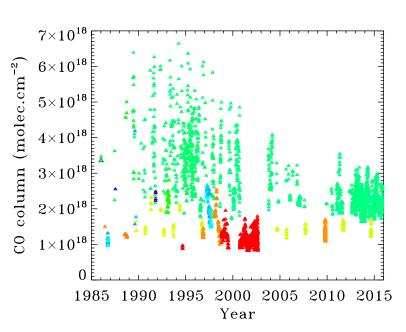


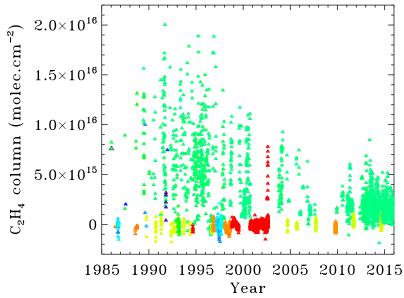
MkIV C₂H₄ / CO correlation

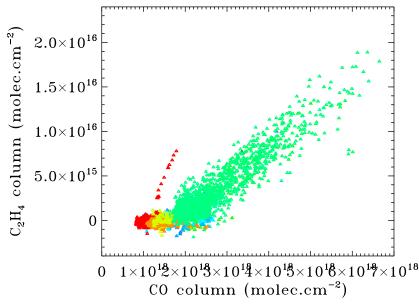
Plot color-coded by altitude, as in previous plot

CO also shows a large decrease over the past 25 years, such that the C_2H_4/CO correlation is very good at JPL.

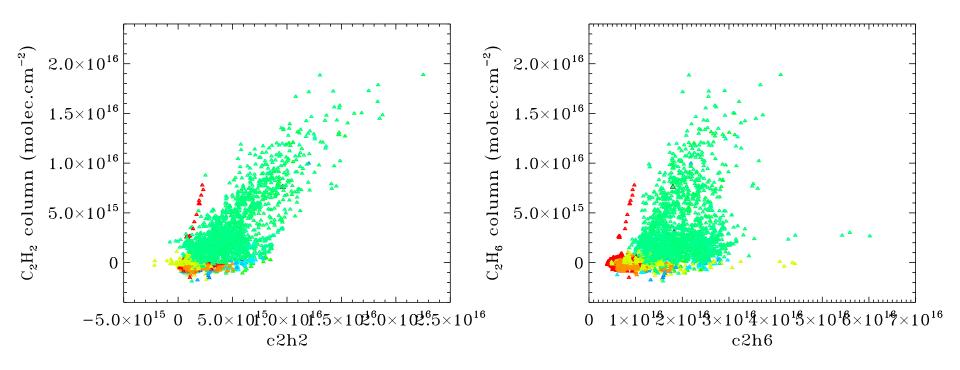
Implies common sources for C₂H₄ and CO, i.e. urban pollution.







C₂H₄ correlations with C₂H₆ and C₂H₂



 C_2H_4 correlations with C_2H_6 and C_2H_2 are positive, but not as good as those with CO.

Comparison with surface in-situ obs

D00V02 WASHENFELDER ET AL.: GLYOXAL BUDGET FOR LOS ANGELES

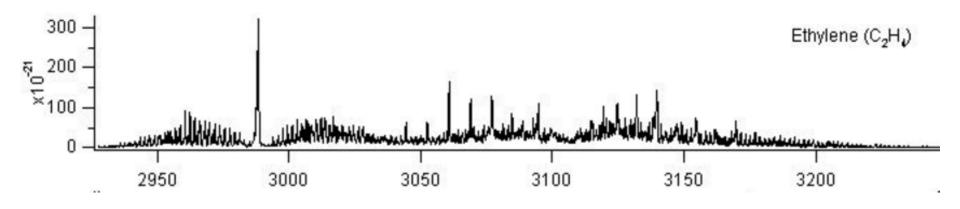
D00V02

Table 2. Master Chemical Mechanism Model Results and Comparison to Measurements

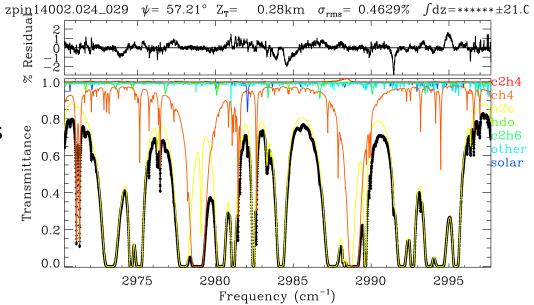
VOC	Initial Concentration (ppbv)	Residual BL Concentration ^a (ppbv)	Emission Ratio VOC/C ₂ H ₂	Measured Concentration ^b (ppbv)	Modeled Concentration ^c (ppbv)	Model-Measured Difference ^d (%)
Ethyne	0.12 ^e	0.24	1.00 ^f	1.78	1.78	0
Ethene	0.065 ^e	0.065	$1.36^{\rm f}$	2.16	1.88	-13
Benzene	$0.039^{\rm e}$	0.040	$0.19^{\rm f}$	0.36	0.34	-4
Toluene	0.017 ^e	0.028	0.82^{f}	0.83	1.19	44
m,p-Xylene	0^{g}	0	$0.37^{\rm f}$	0.32	0.40	26
o-Xylene	0^{g}	0	$0.14^{\rm f}$	0.14	0.17	21
1-Ethylbenzene	0^{g}	0.003	$0.10^{\rm f}$	0.14	0.14	2
n-Propylbenzene	0^{g}	0	$0.028^{\rm f}$	0.022	0.040	82
Isopropylbenzene	0^{g}	0	$0.009^{\rm f}$	0.007	0.013	86
1,2,3-Trimethylbenzene	0^{g}	0	$0.027^{\rm f}$	0.029	0.022	-24
1.2.4 Trimethylbenzene	0^{g}	0	$0.12^{\rm f}$	0.089	0.10	16
Ethene	0^{g}	0.003	N/A	1.09	1.09	0
alpha-Pinene	0^{g}	0	N/A	0.052	0.052	0
Methyl vinyl ketone	0.14^{g}	0.065	1.6×10^{-3h}	0.48	0.41	-14
Methacrolein	0.006^{g}	0.008	4.8×10^{-3h}	0.18	0.24	37
Glyoxal	0	0	6.5×10^{-5h}	0.19	0.18	-5
Acetaldehyde	0.26^{i}	0.5^{j}	0.5^{f}	3.39	2.19	-35

Washenfelder [2011] measured 2.16 ppb of C_2H_4 (Ethene) at ground level in Pasadena, CA. Assuming that this was uniformly distributed throughout the PBL of 150 mbar thickness, the column would be 6×10^{15} molecules/cm². MkIV C_2H_4 columns at JPL in the 2010 timeframe range from 0 to 6×10^{15} and so are consistent.

3000 cm⁻¹ region – hopeless



Measuring C₂H₄ in the 3000 cm⁻¹ region will be difficult. The strongest feature, the nu₁₁ Q-branch at 2988.5 cm⁻¹, is 7 times weaker than the 949 cm⁻¹ feature. Even worse, it coincides with a saturated H₂O line. Perhaps one or two of the Q-branches in the 3060 to 3140 cm⁻¹ region are usable, but these are 15 times weaker than the one at 949 cm⁻¹.



C₂H₄ Summary/Conclusions

C₂H₄ can be measured from sea-level when columns exceed ~2E+15 molec./cm²

I was skeptical at first that C_2H_4 could be retrieved, since its Q-branch is less than 1% deep and is hidden behind a 60% deep CO_2 line that is strongly T-dependent.

But the results are convincing: Under clear, unpolluted conditions C_2H_4 is always zero, whereas in big cities it is highly variable, correlating with CO.

The fact that the fitted window contains 7 other CO_2 lines of similar E" (on average) seems to lessen the impact of CO_2 and T uncertainties on the C_2H_4 retrievals.

 C_2H_4 has a strong correlation with CO, implying similar sources (urban pollution). But C_2H_4 's lifetime is much shorter and so it quickly falls to zero away from sources.

When fitting SF_6 from polluted site, C_2H_4 must be included as an interferent.

No chance of retrieving C_2H_4 from the 3000 cm⁻¹ region, e.g., with a TCCON instrument equipped with an InSb detector.

Aliso Canyon gas leak

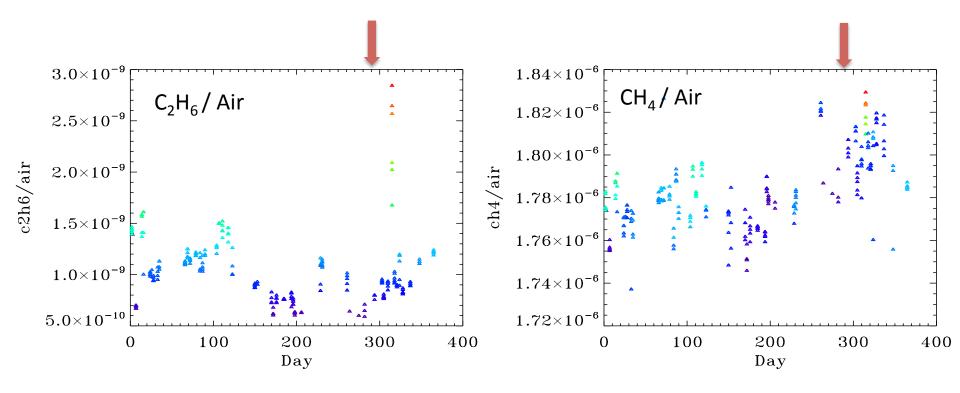
From Wikipedia, the free encyclopedia

The Aliso Canyon gas leak (also called Porter Ranch gas leak^[1] and Porter Ranch gas blowout^[2]) was a massive natural gas leak that was discovered by SoCalGas employees on October 23, 2015.^[3] Gas was released from a well within the Aliso Canyon's underground storage facility in the Santa Susana Mountains near Porter Ranch, Los Angeles.^[4] The second-largest gas storage facility of its kind in the United States, it belongs to the Southern California Gas Company, a subsidiary of Sempra Energy. On January 6, 2016, Governor Jerry Brown issued a state of emergency.^[5] The Aliso gas leak carbon footprint is said to be larger than the Deepwater Horizon leak in the Gulf of Mexico. On February 11, 2016 the gas company reported that it had the leak under control.^[6]

On February 18, 2016, state officials announced that the leak was permanently plugged. An estimated 97,100 tonnes (95,600 long tons; 107,000 short tons) of methane and 7,300 tonnes (7,200 long tons; 8,000 short tons) of ethane was released into the atmosphere, [7] making it the worst natural gas leak in U.S. history in terms of its environmental impact. [8][9][10]

Aliso Canyon Underground Storage Facility is located 40 km from JPL. Jan 4 LA Times states that NG leak began Oct 23, 2015 and peaked on Nov 28 at 60 Tons of CH_4 per hour. By Dec 22 leak rate had decreased to 30 Tons per hour as the underground storage pressure dropped from the initial 2700 psi.

MkIV 2015 C₂H₆ and CH₄ Time Series from JPL

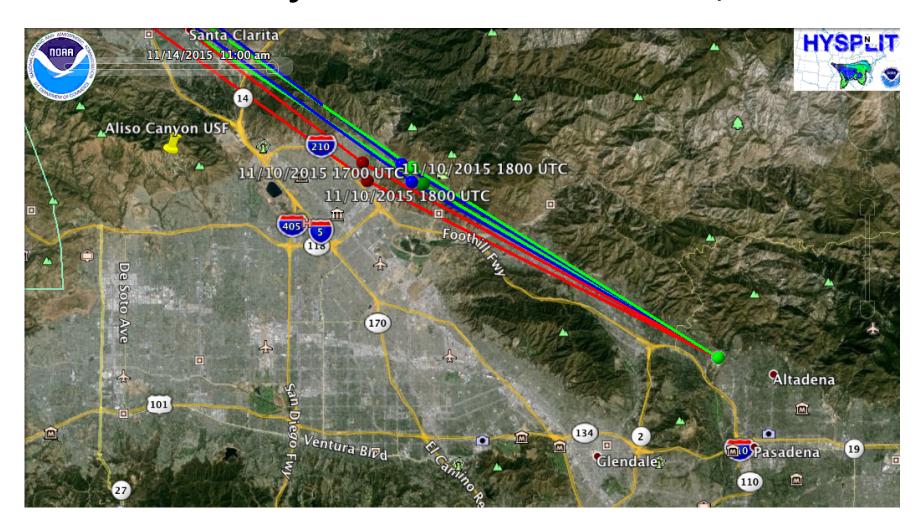


Plots color-coded by C₂H₆ (blue=low; red=high). NG leak began on Oct 23 (red arrow)

Only one day with large C₂H₆ enhancement: by 1.8 ppb (factor 3) on Nov 10

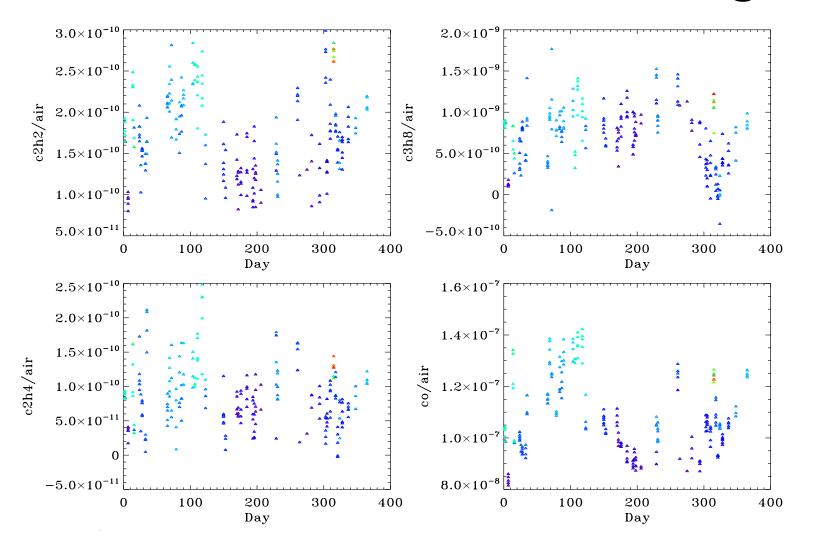
 CH_4 was also slightly high on this day, but enhancement hard to quantify due to natural variations of 1-2%

Back-trajectories for Nov 10, 2015



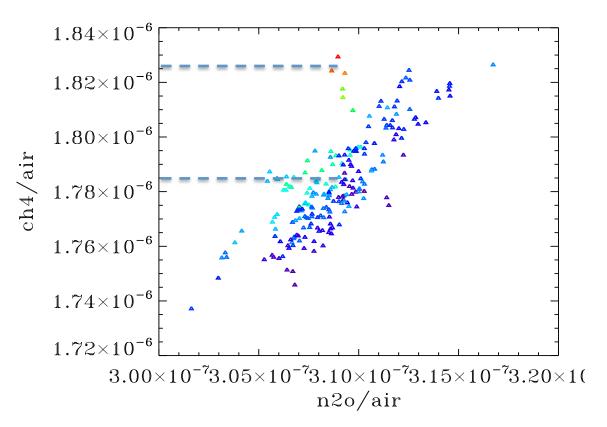
Yellow pin indicates location of Aliso Canyon Underground Storage Facility. Green ball denotes JPL. Within trajectory uncertainties, this represents confirmation that C_2H_6 enhancement on Nov 10 was from Aliso Canyon NG leak.

MkIV 2015 Time Series for other gases



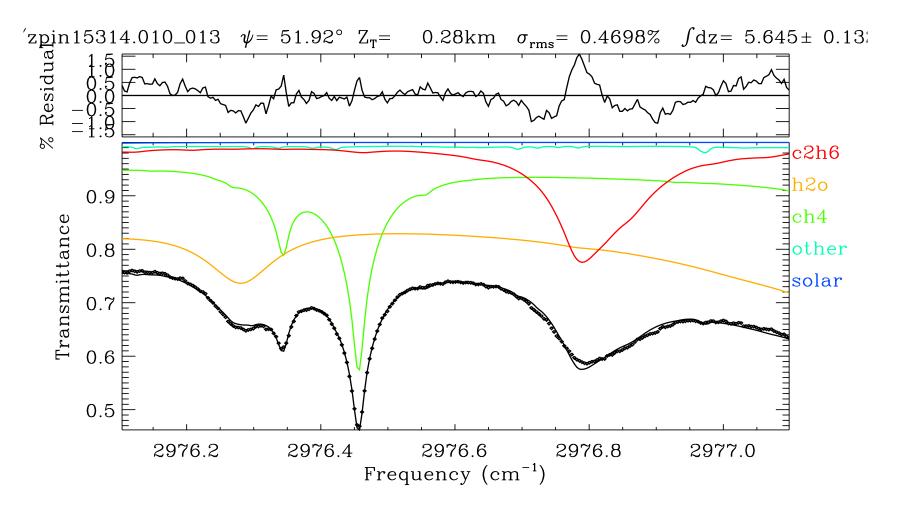
 C_2H_2 , C_2H_4 , C_3H_8 and CO all slightly high on Nov 10, but probably unrelated to NG leak. Plots color-coded by C_2H_6 (blue=low; red=high)

CH₄/N₂O correlations



If no N_2O present in leaked NG, then N_2O can be used to account for dynamics. CH_4 - N_2O correlation shows 0.04 ppm (2.5%) enhancement in CH_4 on Nov 10 (red). 1.8 ppb enhancement of C_2H_6 . Implies NG is 1.8 ppb / 40 ppb = 4.5% C_2H_6 Wikipedia article claims 97,100 tons of CH_4 and 7,300 tons of C_2H_6 , implying leaked NG was 7.0% C_2H_6 by weight or 3.7% by mole fraction.

Spectral fit to C₂H₆ Q-branch on Nov 10



C₂H₆ absorption (red) is under-broadened due to plume being near the surface, not uniformly mixed throughout lower troposphere.

NG Leak -- Summary/Conclusions

Factor 3 enhancement of XC₂H₆ (1.8 ppb) observed from JPL on Nov 10, 2015.

A 2.5% enhancement of XCH₄ (40 ppb) is also seen on this date.

Back-trajectories reveal airmasses passed close to Aliso Canyon, an hour or two before arriving at JPL on Nov 10, 2015.

 C_2H_6 mole fractions inferred from MkIV columns (4.5%) are not inconsistent with composition of leaked NG (3.7%)

Due to the geography, airmasses arriving at JPL from Aliso Canyon have also traversed the densely populated Northern San Fernando Valley. Small enhancements of C_3H_8 , C_2H_4 , C_2H_2 and CO are seen on Nov 10, but are within natural variability.

LA is a large source of C_3H_8 from LPG. In the US in winter, LPG is approximately 90% propane and 10% butane. NG is only ~1% propane.

MkIV column data access

The MkIV data shown in previous slides is available from NDACC-IRWG archive and from the website below.

A single file contains 30 years of data for 28 gases from 12 different sites (mainly Barcroft and JPL).



Ground-Based Observations

Between balloon and aircraft campaigns, the MkIV instrument is used to make ground-based observations. Although these measurements lack the vertical resolution that is a platforms, ground-based observations can nevertheless be made much more frequently - MkIV has averaged over 50 days of observation per year recently. Ground-base accurate method of ascertaining whether the composition of the Earth's atmosphere is changing, which is the main purpose of NDACC.

MkIV Ground-based Vertical Column Abundances 1985-2015

Individual Column Abundances: (89 columns, 4229 rows) m4 avg 1985 2015.vav
Individual Column Abundances: (Ames-1001 format) m4 avg 1985 2015.vav.ames
Daily Average Column Abundances: (89 cols, 1090 rows) m4 avg 1985 2015.vad
List of windows: (265 rows) all mols mir 1985 2015.gnd
Window-to-window biases: m4 avg 1985 2015.vav.cew

Herbin et al. [2009]

GEOPHYSICAL RESEARCH LETTERS, VOL. 36, L04801, doi:10.1029/2008GL036338, 2009



Distributions and seasonal variations of tropospheric ethene (C_2H_4) from Atmospheric Chemistry Experiment (ACE-FTS) solar occultation spectra

H. Herbin, ^{1,2} D. Hurtmans, ¹ L. Clarisse, ¹ S. Turquety, ^{3,4} C. Clerbaux, ^{1,3} C. P. Rinsland, ⁵ C. Boone, ⁶ P. F. Bernath, ^{6,7} and P.-F. Coheur¹

Received 14 October 2008; accepted 8 January 2009; published 17 February 2009.

[1] This work reports the first measurements of ethene (C_2H_4) distributions in the upper troposphere. These are obtained by retrieving vertical profiles from 5 to 20 km from infrared solar occultation spectra recorded in 2005 and 2006 by the Atmospheric Chemistry Experiment-Fourier Transform Spectrometer (ACE-FTS). Background volume

2003; Sawada and Totsuka, 1986]. The dominant mechanism for the removal of C₂H₄ from the atmosphere is the fast reaction with the hydroxyl radical OH [Atkinson et al., 1997]. Secondary sinks include reaction with ozone and transport to the stratosphere [Sawada and Totsuka, 1986]. Because of this rapid reactivity with OH, ethene has a mean

Herbin et al. [2009]

GEOPHYSICAL RESEARCH LETTERS, VOL. 36, L04801, doi:10.1029/2008GL036338, 2009



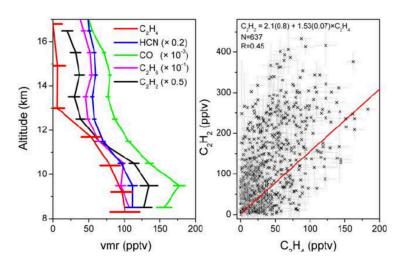
Distributions and seasonal variations of tropospheric ethene (C_2H_4) from Atmospheric Chemistry Experiment (ACE-FTS) solar occultation spectra

H. Herbin, ^{1,2} D. Hurtmans, ¹ L. Clarisse, ¹ S. Turquety, ^{3,4} C. Clerbaux, ^{1,3} C. P. Rinsland, ⁵ C. Boone, ⁶ P. F. Bernath, ^{6,7} and P.-F. Coheur

Received 14 October 2008; accepted 8 January 2009; published 17 February 2009.

 $\ [\]$ This work reports the first measurements of ethene (C_2H_4) distributions in the upper troposphere. These are obtained by retrieving vertical profiles from 5 to 20 km from infrared solar occultation spectra recorded in 2005 and 2006 by the Atmospheric Chemistry Experiment-Fourier Transform Spectrometer (ACE-FTS). Background volume

2003; Sawada and Totsuka, 1986]. The dominant mechanism for the removal of C_2H_4 from the atmosphere is the fast reaction with the hydroxyl radical OH [Atkinson et al., 1997]. Secondary sinks include reaction with ozone and transport to the stratosphere [Sawada and Totsuka, 1986]. Because of this rapid reactivity with OH, ethene has a mean



L04801



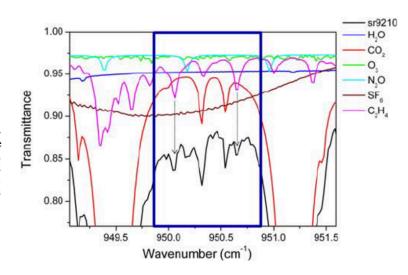


Figure 1. Example of ethene spectral signatures in an ACE-FTS spectrum around 9 km tangent altitude, measured at 26.96°N, 101.79°E on April 28, 2005. The black lirepresents the measured spectrum and the coloured lines the molecule-by-molecule simulations in the spectral window. The dark blue rectangle identifies the spectral microwindow used for the retrievals and the dotted arrows the position of the dominant ethene lines in it.

Figure 4. (left) Example of retrieved C_2H_4 profiles from the same occultation as in Figure 1 and comparison with other carbon tracers. (right) Relationship between the C_2H_4 and C_2H_2 vmrs around 6 km in the troposphere, considering all measured data points. The red curve is obtained by fitting a line through the data points, weighted by their respective uncertainties. The equation of the linear regression, the number of points (N) and the corresponding correlation coefficient. R are given in the upper left corner.

Herbin et al. [2009]

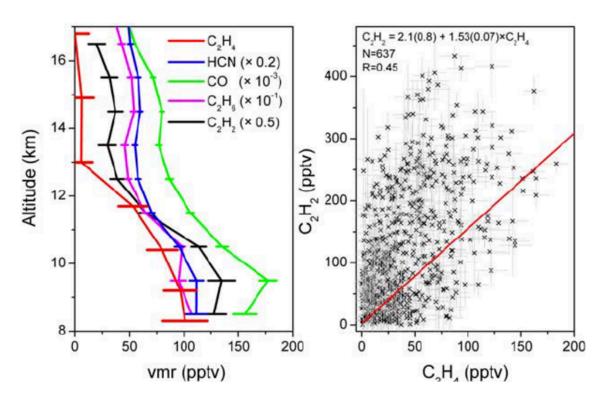


Figure 4. (left) Example of retrieved C₂H₄ profiles from the same occultation as in Figure 1 and comparison with other carbon tracers. (right) Relationship between the C₂H₄ and C₂H₂ vmrs around 6 km in the troposphere, considering all measured data points. The red curve is obtained by fitting a line through the data points, weighted by their respective uncertainties. The equation of the linear regression, the number of points (N) and the corresponding correlation coefficient R are given in the upper left corner.